

PHOSPHACYCLOBUTENES AS INTERMEDIATES IN  
ADDITION OF PHOSPHORANES TO ACETYLENES

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(Received 27 February 1964)

PHOSPHINIMINES (I) react with dimethyl acetylenedicarboxylate to form 1:1 adducts II, the structure of IIb having been proved by X-ray analysis (1) [IIa, m.p. 197-8°, longest  $\lambda_{\max}$  313  $\mu$  ( $\epsilon$  8300); IIb, 180°, 321  $\mu$  (11100); IIc, 243°, 340  $\mu$  (10500); IIId, 235°, 346  $\mu$  (16600); IIe, 244°, 377  $\mu$  (16000)]. Reaction must proceed through the intermediate III, which then breaks the P-N bond. Even the imine IV, which might have undergone Michael addition to give V, reacted in the same way to give an adduct [m.p. 170°, longest  $\lambda_{\max}$  289  $\mu$  ( $\epsilon$  5000)] with structure VI, assigned on the basis of the NH band at 3380  $\text{cm}^{-1}$  and the similarity of its spectra and those of its salts with those of IIa and its salts. When the nitrogen atom was made less nucleophilic than in Ie, as in the *N*-2,4-dinitrophenyl-, -benzoyl-, -ethoxycarbonyl- or -*p*-toluenesulphonyl-imines addition to the acetylene ester no longer occurred. VII, R= Ph, C<sub>6</sub>H<sub>5</sub> and *p*-SO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub> also failed to react.

Phosphoranes with no  $\alpha$ -hydrogen atom added to dimethyl acetylenedicarboxylate in the same way. Thus pentaphenylphosphorane (VIII) gave IX [m.p. 257°, longest  $\lambda_{\max}$  365  $\mu$  ( $\epsilon$  3000) presumably via X, and the fluorenylidene analogue gave a similar adduct [m.p. 211°, 427  $\mu$  (10000)]. Hendrickson (2) has just described the addition of C-benzoyl-methylene-triphenylphosphorane to dimethyl acetylene-

dicarboxylate both by Michael addition and through the phosphacyclobutene.

In methylene chloride or dry ether at 0° the phosphazines (XI) gave adducts with dimethyl acetylenedicarboxylate [XIIa, m.p. 126-7°,  $\lambda_{\max}$  368 m $\mu$  (13800); XIIb, 140°, 366 m $\mu$  (14000)] formed by Michael addition. In warm ethyl acetate these adducts (XII) decomposed to triphenylphosphine and the pyrazole-triester (XIII). The single proton that appears at  $\tau$  3.45 (CDCl<sub>3</sub> solution) in the N.M.R. spectrum of XIIa is evidence against addition to the P=N bond (the parent phosphazine, XIa, has no olefinic proton at higher field than the aromatic ones). On the other hand the phosphazine XIV, with no  $\alpha$ -hydrogen atoms, added via the phosphazacyclobutene to give the adduct XV [m.p. 225°, 417 m (18000)].

This facile opening of the phosphacyclobutene and phosphazacyclobutene rings is reminiscent of the isomerisation of the cyclobutene systems (3) from addition of dimethyl acetylenedicarboxylate to enamines.

G.W.B. acknowledges a D.S.I.R. Research Studentship.

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